Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

## Shu-Ping Yang,<sup>a</sup>\* Li-Jun Han<sup>b</sup> and Da-Qi Wang<sup>c</sup>

<sup>a</sup>Department of Chemical Engineering, Huaihai Institute of Technology, Lianyungang 222005, People's Republic of China, <sup>b</sup>Department of Mathematics and Science, Huaihai Institute of Technology, Lianyungang 222005, People's Republic of China, and <sup>c</sup>College of Chemistry and Chemical Engineering, Liaocheng University, Shandong 252059, People's Republic of China

Correspondence e-mail: yangshuping@hhit.edu.cn

#### **Key indicators**

Single-crystal X-ray study T = 298 KMean  $\sigma$ (C–C) = 0.005 Å R factor = 0.045 wR factor = 0.124 Data-to-parameter ratio = 12.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 4-Methyl-7-phenylsulfonamido-2*H*-1-benzopyran-2-one

In the title compound,  $C_{12}H_{11}NO_3$ , the molecules have two intramolecular  $C-H\cdots O$  hydrogen bonds that define an S(6)S(5) pattern. The molecules are linked by  $N-H\cdots O$ hydrogen bonds into a centrosymmetric dimer. These dimers are linked by further  $C-H\cdots O$  hydrogen bonds, so forming a [100] ribbon of  $R_2^2(8)R_4^4(18)[R_4^4(20)]$  rings, a [111] chain of alternating  $R_2^2(8)$  and  $R_2^2(22)$  rings, and an [010] complex chain of  $R_4^2(18)$  rings. The combination of the [100] ribbon, the [111] chain and the [010] chain results in the formation of a threedimensional network structure.

## Comment

Aminocoumarin derivatives are known to be a category of important fluorogenic dyes (Lemieux *et al.*, 2003; Zhao *et al.*, 2004). Some aminocoumarin derivatives show high activity as antiplatelet agents (Roma *et al.*, 2003). We recently reported the molecular structure of an aminocoumarin (Yang *et al.*, 2006). Here, we report the molecular structure of the title aminocoumarin derivative, (I).



In the molecule of (I), the molecule adopts a similar 'vault' conformation, and two intramolecular  $C-H \cdots O$  hydrogen bonds define an S(6)S(5) pattern (García-Báez *et al.*, 2002) (Fig. 1 and Table 1). The dihedral angle between the sixmembered ring O3/S1/N1/C7/C6/H1 and the five-membered ring O3/S1/C11/C12/H12 is 80.28 (9)°. The coumarin unit and the phenyl ring (C11–C16) make a dihedral angle of 75.21 (8)°. The geometric parameters for (I) are normal (Allen *et al.*, 1987).

In the crystal structure of (I), the molecules are linked by a pair of  $N-H\cdots O$  hydrogen bonds into a centrosymmetric dimer of  $R_2^2(8)$  ring motif (Bernstein *et al.*, 1995), centred at  $(\frac{1}{2}, 1, 1)$  (Fig. 2 and Table 1).

Received 18 November 2006 Accepted 27 November 2006

All rights reserved

© 2007 International Union of Crystallography



## Figure 1

The molecular structure of compound (I), showing the atom-labelling scheme and intramolecular hydrogen-bonded S(5)S(6) rings. Displacement ellipsoids are drawn at the 30% probability level. Dashed lines indicate hydrogen bonds.



## Figure 2

Part of the crystal structure of (I), showing the formation of a dimer of  $R_2^2(8)$  ring motif. For clarity, H atoms not involved in the motif shown have been omitted. Dashed lines indicate hydrogen bonds [Symmetry code: (\*) 1 - x, 2 - y, 2 - z].

These dimers are the basic building block in the formation of three one-dimensional substructures. Atom C9 in the molecule at (x, y, z) acts as hydrogen-bond donor to atom O3 in the molecule at (-1 + x, y, z), so forming a centrosymmetric  $R_4^4(18)$  ring centred at (0, 1, 1). These dimers are linked by C-H···O hydrogen bonds into a ribbon of  $R_2^2(8)R_4^4(18)[R_4^4(20)]$ rings (García-Báez et al., 2002) along the [100] direction (Fig. 3 and Table 1). Atom C12 in the molecule at (x, y, z) acts as hydrogen-bond donor to atom O2 in the molecule at (2 - x,1 - y, 1 - z); propagation of the C-H···O interaction by inversion, generates a centrosymmetric  $R_2^2(22)$  ring centred at  $(1, \frac{1}{2}, \frac{1}{2})$ , and these dimers are linked into a chain of alternating  $R_2^2(8)$  and  $R_2^2(22)$  rings along the [111] direction (Fig. 4). Atom C14 in the molecules at (x, y, z) and (1 - x, 1 - y, 2 - z) act as hydrogen-bond donors to atoms O4 in the molecules at (x, x)-1 + y, z) and (1 - x, 2 - y, 2 - z), respectively, linking adjacent dimers into a complex [010] chain of rings and forming a centrosymmetric  $R_4^2(18)$  ring centred at  $(\frac{1}{2}, \frac{3}{2}, 1)$ (Bernstein et al., 1995), also linking the dimers into a complex





Part of the crystal structure of (I), showing the formation of a ribbon of  $R_2^2(8)R_4^4(18)[R_4^4(20)]$  rings along the [100] direction. For clarity, H atoms not involved in the motif shown have been omitted. Dashed lines indicate hydrogen bonds. [Symmetry codes: (\*) 1 - x, 2 - y, 2 - z; (#) -1 + x, y, z; (&) -x, 2 - y, 2 - z; (\$) 2 - x, 2 - y, 2 - z; (@) 1 + x, y, z].

[010] chain of rings (Fig. 5). The combination of the [010] chain, the  $[1\overline{11}]$  chain and the [100] ribbon results in the formation of a three-dimensional network structure.

## **Experimental**

To a solution containing 7-amino-4-methylcoumarin (1.75 g, 10 mmol) and anhydrous pyridine (10 ml), a solution of phenylsulfonyl chloride (11 mmol) was slowly added at 278–283 K with stirring over a period of 30 min. The reaction mixture was then stirred continuously for 12 h at room temperature (298–300 K) and then poured into ice–water (200 ml). The solid obtained was filtered off, washed with water and dried at room temperature. Yellow crystals of (I) suitable for X-ray structure analysis were obtained by evaporation of an ethanol solution of the product over a period of one week (m.p. 518–519 K).

Crystal data	
$C_{16}H_{13}NO_4S$	V = 717.0 (5) Å <sup>3</sup>
$M_r = 315.33$	Z = 2
Triclinic, P1	$D_x = 1.461 \text{ Mg m}^{-3}$
a = 8.040 (3)  Å	Mo $K\alpha$ radiation
b = 8.193 (3) Å	$\mu = 0.24 \text{ mm}^{-1}$
c = 11.656 (5)  Å	T = 298 (2) K
$\alpha = 83.246 \ (5)^{\circ}$	Block, yellow
$\beta = 85.316 \ (5)^{\circ}$	$0.42 \times 0.25 \times 0.21 \text{ mm}$
$\gamma = 70.287 \ (4)^{\circ}$	

## Data collection

Siemens SMART 1000 CCD areadetector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan

(SADABS; Sheldrick, 1996) $T_{min} = 0.905, T_{max} = 0.951$ 



## Figure 4

Part of the crystal structure of (I), showing the formation of a chain of alternating  $R_2^2(8)$  and  $R_2^2(22)$  rings along the [111] direction. For clarity, H atoms not involved in the motif shown have been omitted. Dashed lines indicate hydrogen bonds. [Symmetry codes: (\*) 1 - x, 2 - y, 2 - z; (#) 2 - x, 1 - y, 1 - z; (&) 1 + x, -1 + y, -1 + z; (\$) -1 + x, 1 + y, 1 + z; (@) -x, 3 - y, 3 - z].

Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0566P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.046$	+ 0.2626P]
$wR(F^2) = 0.124$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.05	$(\Delta/\sigma)_{\rm max} < 0.001$
2493 reflections	$\Delta \rho_{\rm max} = 0.24 \text{ e } \text{\AA}^{-3}$
200 parameters	$\Delta \rho_{\rm min} = -0.34 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

### Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
С6-Н6О3	0.93	2.46	3.099 (4)	126
$N1-H1\cdots O4^{i}$	0.90	2.07	2.923 (3)	157
C9−H9···O3 <sup>ii</sup>	0.93	2.51	3.287 (3)	141
C12-H12···O2 <sup>iii</sup>	0.93	2.72	3.278 (4)	120
$C14-H14\cdots O4^{iv}$	0.93	2.57	3.336 (4)	140
Symmetry codes:	(i) $-x +$	1, -y + 2, -z +	+2; (ii) $x -$	-1, y, z; (iii)

-x + 2, -y + 1, -z + 1; (iv) x, y - 1, z. (ii) x + 1, y + 2, z + 2, (ii)

All H atoms were positioned geometrically and refined as riding on their parent atoms, with C-H = 0.96 Å and  $U_{iso}(H)$ = 1.5 $U_{eq}(C)$ for methyl H atoms, N-H = 0.90 Å and  $U_{iso}(H)$ = 1.2 $U_{eq}(C)$  for amino H atoms, and C-H = 0.93 Å and  $U_{iso}(H)$  = 1.2 $U_{eq}(C)$  for all other H atoms.



#### Figure 5

Part of the crystal structure of (I), showing the formation of a chain of hydrogen-bonded dimers along [010]. For clarity, H atoms not involved in the motif shown have been omitted. Dashed lines indicate hydrogen bonds. [Symmetry codes: (\*) 1 - x, 2 - y, 2 - z; (#) x, -1 + y, z; (&) 1 - x, 1 - y, 2 - z; (\$) 1 - x, 3 - y, 2 - z; (@) x, 1 + y, z].

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

The authors acknowledge the financial support of the Huaihai Institute of Technology Science Foundation.

### References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1573.
- García-Báez, E. V., Martínez-Martínez, F. J., Höpfl, H. & Padilla-Martínez, I. I. (2002). Cryst. Growth Des. 3, 34–45.
- Lemieux, G. A., de Graffenried, C. L. & Bertozzi, C. R. (2003). J. Am. Chem. Soc. 125, 4708–4709.
- Roma, G., Di Braccio, M., Carrieri, A., Grossi, G., Leoncini, G., Signorello, M. G. & Carotti, A. (2003). *Bioorg. Med. Chem.* 11, 123–138.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Yang, S.-P., Han, L.-J., Wang, D.-Q. & Ding, T.-Z. (2006). Acta Cryst. E62, 05196–05198.
- Zhao, Y. R., Zheng, Q., Dakin, K., Xu, K., Martinez, M. L. & Li, W. H. (2004). J. Am. Chem. Soc. 126, 4653–4663.